

## Triarylbismuth as green crosscoupling reagent : experimental versus theoretical mecanism insights

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Triarylbismuth are stable and easily prepared orgametallic compounds. [1] Their main features come from this low toxic heavy metal ( $^{209}$ Bi), trivalent with a slightly basic lone pain (*p*Ka ~5.2). This organometallics are efficient partners in metal-catalyzed crosscoupling reactions, mainly with copper or better with palladium where the 3 aryles are effectively transferred. [2] However this last reaction suffer from one side reaction : the dimer formation, which can be lowered using PEPPSI. [3]



The efficiency of these reactions is strongly dependent on the arylsusbtituants: electronic properties and relative position toward Bi atom. Furthermore, PEPPSI is robust enough to allow domino reaction involving an elimination/coupling/coupling process as described below.



A structure reactivity relationship study led from an experimental and theoretical point of view (DFT) pointed out a pretty classical Suzuky type mechanism for the aryl transfer, where the active species is and NHC-Pd-PPh<sub>3</sub>. The rate determining step is the transmetallation's one which is even slower with an Ar<sub>3</sub>Bi than an ArBiX than an ArBiX<sub>2</sub>. Interestingly intramolecular chelation favor this step. [4,5]

References :

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